Notes – Particles in Motion

Particles in Motion	All is made of tiny
Particles in Motion	These particles are in constant all forms of matter to have different
Phases of Matter	Phases (States) of Matter
	Liquid
Solids	Particles are packed, close together and moving Definite and because particles are packed close
Liquids	Particles slowly one another definite but do have a definite volume
Gases	Particles are very apart and move very definite and definite
Plasma	are extremely far and
	Basically, is a hot gas that gives off
	Examples, and halogen lights in classroom

Phase	to a
Changes	to ato a
0	to a
	to a
Phase	change from a to a
Changes	The best known example is "", solid CO ₂
8	when a substance changes directly from a
	to a
	Example-formation of
Changes in Density	is the ratio of per unit of a substance.
	The means how much is in the substance
	the is the the matter takes up.
	The more packed the particles are in a certain space, the the
Changes in Density	 Almost all are more solids than a
Exception: Ice & Water	Because of the way bonds, when water freezes to ice, it leaves between the molecules.
	These spaces make the ice (solid form of water) dense.