| Physical Change | A | Change is when an | object | |
|---------------------|-----------------------------------------------------------------------|---------------------------------------|------------------------------------------------------------|--|
| | a physical | | | |
| | A change in | , shape, | , or | |
| | · | | | |
| | The object | is the | thing. | |
| | Examples of Physical Chang | ges | | |
| | | Paper | | |
| | | Water | | |
| | | Hair | | |
| | | Brick | | |
| Chemical Change | A | | | |
| | the chemical of the object. | | | |
| | The object | a | object. | |
| | Something | is | | |
| Keys to Identifying | 1 | change | | |
| a Chemical Change | | | | |
| | 3. Heat, | _ or Sound is given off. | | |
| | 4. Formation of a | | | |
| | or | | | |
| | 5. Formation of a | | | |
| | | | | |
| | A new colour Heat, light, or s appears. is given of (or absorbe | f are formed. A (called a precipitate | The change is e) difficult or impossible to reverse. | |

| Chemical Change | Examples of Chemical Changes | |
|--------------------|------------------------------------------------------------------------|--|
| 3 | • exploding | |
| | • An apple | |
| | | |
| | • cookies | |
| | Photosynthesis/Respiration | |
| | • | |
| | • Combustion () | |
| Chemical Reactions | | |
| Chemical Reactions | Chemical Reaction Ain which some substances into different substances. | |
| | Ex: Magnesium + → Magnesium Oxide | |
| | | |
| Chemical Equations | starting substance | |
| | – starting substanceOn of arrow | |
| | Products – substance what is | |
| | | |
| | • On of arrow | |
| | • → Products | |
| | | |
| | | |
| | Reactants Products | |
| | $4 \text{ Fe} + \text{O}_2 \longrightarrow 2 \text{ Fe}_2 \text{O}_3$ | |
| | 21023 | |
| | | |
| | | |