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LAB: GEL WORMS

Background:

Sodium alginate is made from a type of brown seaweed called kelp which grows wild in the ocean. It is harvested and processed to make sodium alginate. Alginate is used in food because it is a powerful thickening and gel-forming agent. Some foods that may include alginate are ice cream, fruit snacks, salad dressings, pudding and onion rings.



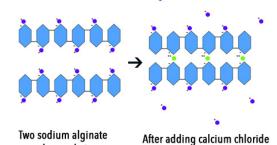
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Calcium chloride is made from limestone which is a common rock that is mined. The limestone is reacted with hydrochloric acid or sodium chloride to make the calcium chloride.



In order to form a gel, sodium alginate needs to come in contact with calcium ions. When sodium alginate reacts with calcium chloride, a polymer is formed (which is a long chain molecule). This polymer is a gel.

<u>Materials</u>: plastic container, syringe, sodium alginate solution, calcium chloride solution



polymer chains

Directions:

- 1. Pour 200 mL calcium chloride solution into the plastic container.
- 2. Fill the syringe with 25 mL sodium alginate solution.
- 3. Using the syringe, add the sodium alginate solution to the calcium chloride solution in the plastic container.... Add one drop at a time to get beads or squeeze the solution into the beaker to make strings or worms.

Clean up:

- 1. Get the "worms" out of the calcium chloride solution and throw in the trash.
- 2. Save the calcium chloride solution for the next class.
- 3. Wash the syringe with hot water and leave open to dry.

Questions:

- 1. Is this experiment a physical or chemical change?
- 2. What natural substance does sodium alginate come from?
- 3. Why is sodium alginate used in food?
- 4. What natural product does calcium chloride come from?
- 5. What is formed when sodium alginate reacts with calcium chloride?

Notes: Natural/Synthetic

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SyntheticMade	
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Examples of Synthetic Materials: (and the natural resou	urce they are made from)
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